

Chemical Phosphorus Separation

Experiences from Michigan Dairy Farms

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Why consider phosphorus separation?



Background

- Chemical precipitation makes phosphorus insoluble
 - Algae need *soluble nutrients*
- Chemical coagulants used in water and wastewater treatment since 1970's
 - Driven by the Clean Water Act and the need for more complete nutrient removal

Waste strength: Dairy vs. Municipal

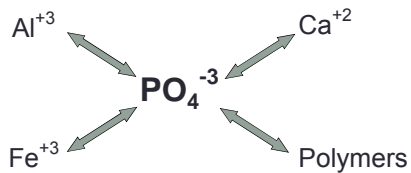
Parameter	Municipal Wastewater (mg/L)	Dairy Manure (mg/L)
Color	Grey to black	Brown to black
BOD5	250	15,000
COD	500	37,000
Total N	40	6,000
Total P	9	3,000
Total solids	300	95,000
Volatile solids	240	81,000

Phosphorus separation options

- Biological
 - Cycling of anaerobic and aerobic conditions
- Chemical precipitation
 - Addition of chemical compounds to bind soluble phosphorus
- Struvite formation (magnesium ammonium phosphate)
 - Crystallized growth containing phosphate, ammonium, and magnesium

Chemical phosphorus separation

- Chemical phosphorus removal is based on the principles of general chemistry



"OPPOSITE CHARGES ATTRACT"

Terminology

- Precipitation & coagulation
 - Change from a liquid to thickened, curdlike, insoluble state
 - Brought about by the addition of chemical coagulants
 - Charge neutralization brought about by interaction
 - Destabilization of colloids by the addition of coagulants
- Flocculation
 - Aggregation of small particles into larger particles
 - Interaction of destabilized particles to form dense flocs
 - Aided by the addition of polymers
 - Excessive mixing may breakdown floc
- Separation
 - Passive
 - Mechanically
 - Belt filter press or dissolved air floatation

Factors impacting chemical selection

- Influent phosphorus
- Desired phosphorus reduction
- Wastewater suspended solids
- Alkalinity
- Chemical cost
- Reliability of chemical supply
- Climatic conditions
- Sludge handling facilities
- Ultimate disposal method
- Compatibility with other treatment processes
- Changes in feed ration

Metcalfe and Eddy, Wastewater Engineering Treatment, Disposal, and Reuse, 1991

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Equipment

- Rapid mix chamber with chemical injector
- Flocculation tank
- Separation devices
 - Belt filter press
 - Dissolved air floatation (DAF)
 - Clarifier
 - Screw press
 - Settling cell

Equipment



Experience from Michigan

- Three systems installed in early 2000's
 - Green Meadow Farms (2002)
 - Belt press
 - 3,300 cows (sand)
 - Maple Row Dairy Farm (2004)
 - Belt press
 - 2,500 cows and replacements (sand)
 - Bradford Dairy (2003)
 - Dissolved air floatation
 - 1,200 cows (organic bedding)
- Driven largely by regulatory uncertainty
 - NPDES permits new
 - Michigan Right to Farm law (GAAMP's)

Experience from Michigan

Reason for system adoption

- Green Meadow Farms
 - Existing irrigation system
 - Manure management labor
- Maple Row Dairy Farm
 - Concern over manure application
 - Wanted to install irrigation
 - Reduce manure transport
 - Improved crop yield
- Bradford Dairy
 - Proximity to the city of Sparta
 - Manure management labor

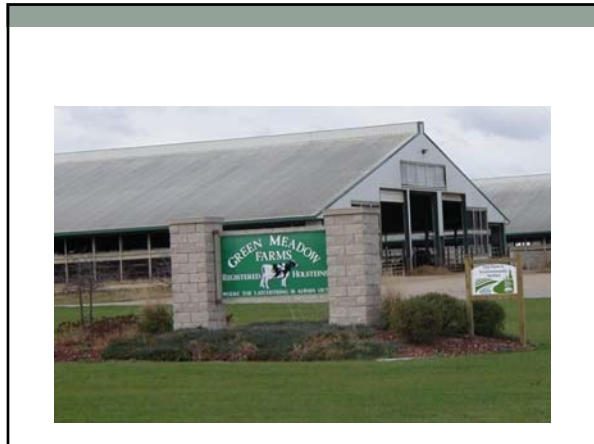
General phosphorus system goals

- Effluent characteristics
 - >90% soluble phosphorus reduction
 - >50% solids reduction
 - Low odor
- Separated solids - "cake"
 - >20% total solids
 - Concentration of phosphorus, organic nitrogen and solids
 - Compostable
- Chemical
 - Minimal usage
 - Cost effective

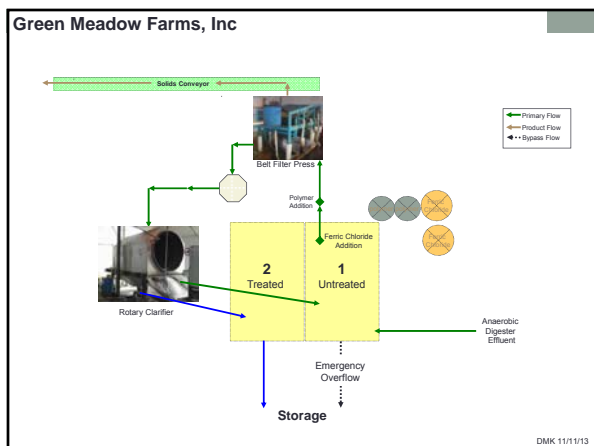
Experience from Michigan

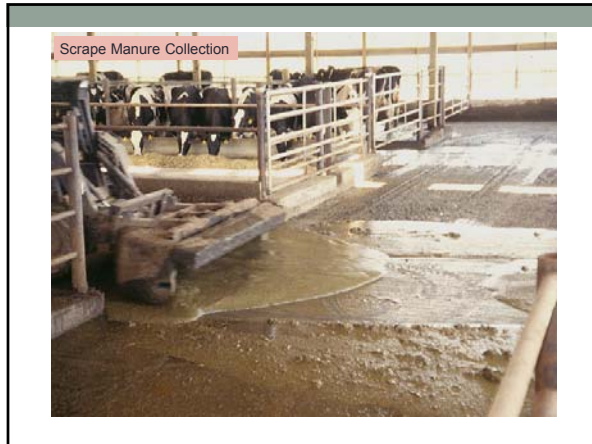
System specifics

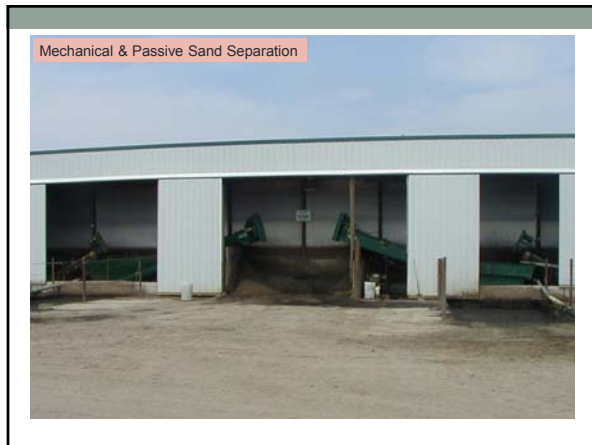
- Green Meadow Farms
 - Sand separation at the source
 - Pump to manure treatment center
 - Chemical cost ≈ \$73/yr/cow (2003)
- Maple Row Dairy Farm
 - Vacuum manure from the barn alley, transport to manure treatment center
 - Entire manure processing system under one roof
- Bradford Dairy
 - Gravity flow to manure treatment center
 - Entire manure processing system under one roof
 - Chemical cost ≈ \$50/yr/cow (2003)

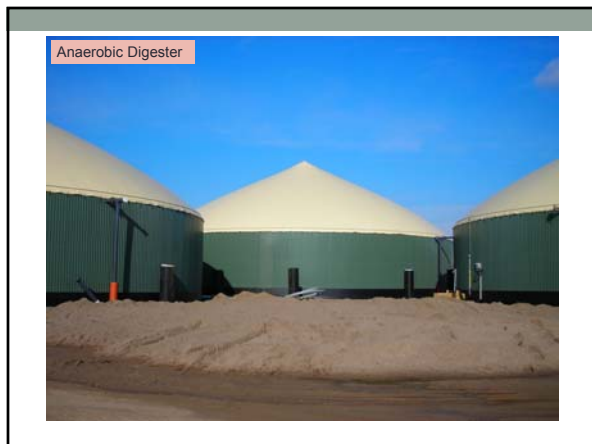












Chemical Phosphorus Separation System



Chemical Storage



Inline Ferric Mixing





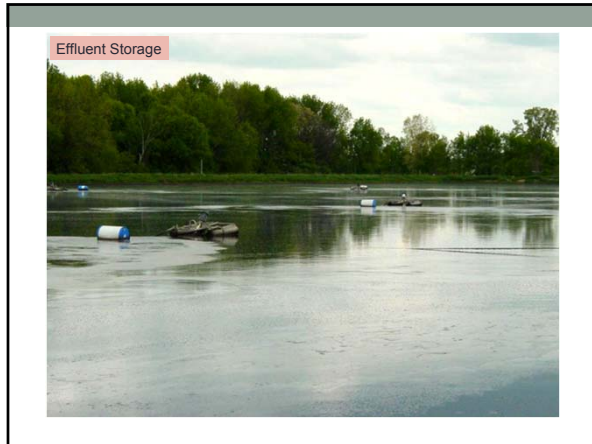


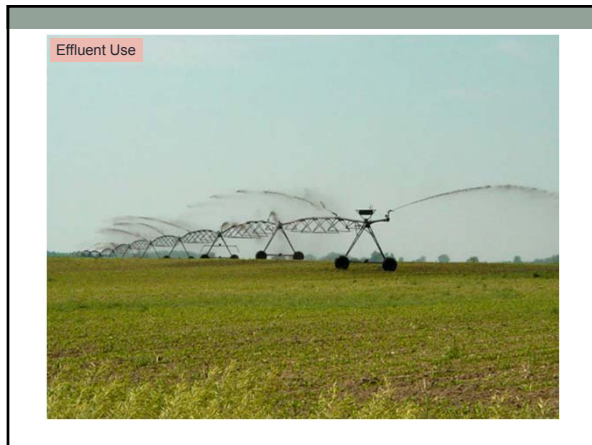












System products

Parameters	Input	Effluent	Solids	
	(lb/1,000 gal)	(lb/1000 gal)	(lb/ton)	(lb/1000 gal)
Total Solids	4.2%	0.4%	16.6%	16.6%
Total Nitrogen	14.5	5.3	10.9	45.5
Ammonia (NH ₄ -N)	7.2	4.6	1.70	7.1
Phosphorus (P ₂ O ₅)	5.7	0.2	5.2	21.7
Potassium (K ₂ O)	11.3	7.7	2.8	11.7

Treatment costs (chemicals)

- Flow rate – 110,000 to 120,000 gpd

Chemical	Usage (lb/d)	Cost
Ferric chloride	16,000	\$1,150
Organic polymer	230	\$360
De-foamer	40	\$30

- Approximate treatment cost per gallon \$0.013
- MI application cost \$0.01 to \$0.032 per gallon

Technology challenges

- Capital investment
- Chemical cost
- Maintenance
- Engineering is of paramount concern
- System reliability
 - Step learning curve
 - Some redundancy needed
- Fate of bound phosphorus and chemical additives
 - Stability of phosphorus complex
 - Addition of metals to cropland

Benefits of chemical separation

- >90% reduction in soluble phosphorus
- Reduction in fresh water consumption
 - Less to dispose
- Marketable commodity
- Odor reduction
- Improved nutrient management
 - Time, form, rate, & placement

Conclusions

- Innovative technologies are here...
 - More research is needed
 - Experience will come from those who take the risk
- Wide range of options
 - Nutrient separation ranges from minimal to near 100%
 - Simple to intensive
 - Inexpensive to \$\$\$\$\$
- Anaerobic digestion is beneficial for pretreatment
 - Stabilized pH, temperature & flow
 - Seasonal changes do still occur

Questions

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